



## REQUEST FOR PROPOSALS (RFP)

### *Assessing Changing Salinity in Water Sources (5308)*

#### **Date Posted**

Monday, September 9, 2024

#### **Due Date**

Proposals must be received by 3:00 pm Mountain Time on Thursday, November 14, 2024.

#### **WRF Project Contact**

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#### **Project Sponsors**

This project is funded by The Water Research Foundation (WRF) as part of WRF's Research Priority Program.

#### **Project Objectives**

- Conduct a regional assessment of salinity impacts to drinking water sources, treatment processes (e.g., high versus low pressure membranes), and water reuse applications to produce a One Water categorical overview of risk types.
- Provide a review of current real-time measurement parameters, research, and assessments by national and international organizations (e.g., United States Geological Survey (USGS)) as well as river commissions and regional working groups (e.g., Delaware River Basin Commission, South Platte Salinity Stakeholder Group, Southern California Salinity Coalition, Adirondack Road Salt Reduction Task Force, etc.) based on the risk types identified in the objective above, including utility-specific case studies and investigations.
- Understand the sources of salinity and their relative contributions to drinking water sources via case study investigations or modeling.
- Develop communication materials for use by the water sector to educate the public—including customers, staff, and other community stakeholders—about the risks of increased salinity and efforts recommended to reduce the problem.

#### **Budget**

Applicants may request up to \$150,000 in WRF funds for this project.

#### **Background and Project Rationale**

Conventional drinking water treatment does not remove total dissolved solids (TDS) from water supplies. In water supplies that have changing and increasing TDS, finished water quality,

ecological and aquatic health, public health, and water infrastructure may be impacted. There are multiple sources of supply considered at risk of having increased or increasing TDS, specifically salinity. Contributing sources or activities that may increase salinity, the amount of salt dissolved in water, include:

- Growing populations and increased water demand
- Seasonal usage of brine and road salt applications for de-icing that impact surface and ground water supplies
- Flowback water from hydraulic fracturing used for natural gas extraction
- Seawater intrusion from over-pumping/overuse of groundwater and impacted aquifers
- Exchanges between surface waters and groundwater systems
- Changes in tidal activity (sea level rise or other climate impacts) for waterways that are tidally influenced (e.g., Delaware River)
- Drought
- Municipal and industrial wastewater discharges (i.e., use of household softeners, water treatment chemical additions)
- Naturally occurring geological sources
- Agricultural practices

Freshwater salinization affects water supplies directly through a measurable increase in sodium and chloride levels, and indirectly by increasing stress and/or mortality of freshwater organisms, thereby resulting in ecosystems losing their ability to provide clean water. In addition, TDS levels can also impact finished water taste, and certain components of TDS (e.g., chlorides, sulfates, magnesium, calcium, carbonates) also cause corrosion or encrustation in water distribution systems. Specific conductance is frequently used as a surrogate for salinity and TDS. The United States Environmental Protection Agency (USEPA) lists TDS as a Secondary Drinking Water Standard and recommends treatment when concentrations exceed 500 mg/L. There are different state or local TDS limits for drinking water (e.g., Nevada's Secondary Drinking Water Standard for TDS is 1,000 mg/L) and wastewater in which the drivers for the selection of specific numerical values of the respective standards should be summarized. USEPA also has standards for chloride and sodium.

In addition to the secondary standard for TDS, other potential adverse impacts of increased salinity include increases in disinfection byproduct (DBP) formation and an observed shift of DBP concentrations and types to more toxic analogues, such as brominated and iodinated forms. Public health research continues to delineate exposure routes through drinking water for persons in coastal populations, specifically. Other brominated and DBP precursor contributions to salinity can come from energy production (e.g., power plants and fracking). Reducing brominated DBPs through bromide control has been discussed; one approach is to control upstream dischargers (Good and VanBriesen 2017). Numerous studies have demonstrated increases in brominated and iodinated DBPs (e.g., trihalomethanes, haloacetic acids, haloacetonitriles) (Fellet 2014, Parker et al. 2014). These DBP precursors and types are critical concerns for public health since brominated and iodinated analogues are more cyto- and

genotoxic. The water sector is anticipating regulatory changes with the upcoming Microbial/Disinfection Byproducts Rule revisions.

Investigations that underscore individual challenges and supply management include the case study in Brick Township, NJ, in which a treatment plant switched their supply from river to reservoir when specific conductivity increased (Rosenberg et al. 2022). Increases from urbanization and de-icing requirements have been studied in the eastern US. Understanding the relative contribution of chloride from big-box store parking lots and sidewalks to state/municipal Departments of Transportation (DOTs) would be valuable.

Further, utilities want to better understand how long it will take for salinized aquifer sources to recover after adopting best management practices for de-icing. Would chloride continue to increase for some time before decreasing? The Northeast and Mid-Atlantic regions are subject to winter snowfall and ice. New Hampshire evaluated the impact of road salt on their drinking water wells and has listed 50 chloride-impaired water bodies (NHDES 2021). New York has leveraged geographic information system (GIS) tracking of sodium and chloride concentrations in their aquifers to determine impacted areas (Kelly et al. 2018). In addition to increases in salinity, other adverse outcomes can occur, such as the mobilization of other contaminants (i.e., radon, heavy metals) and increased corrosion potential.

While salinity threatens the quality of drinking water sources in certain parts of North America, water resource recovery facilities (WRRFs) are also increasingly challenged with meeting discharge limits for chloride, sulfate, iron, and/or aluminum. These limits are set by regulators to protect drinking water source quality or other uses. As an example, the Colorado Department of Public Health and Environment (CDPHE) proposed limits for chloride and sulfate driven by antidegradation protection that are well below secondary drinking water standards. Compliance is becoming increasingly difficult as Colorado dischargers are also required to add metals salt to meet progressively lower nutrient standards.

The build-up of salinity in surface and groundwaters over time poses risks to public and aquatic life and challenges to the water sector as a whole, including water resource management, drinking water providers, WRRFs, and potable and non-potable water reuse systems. Salts are challenging to remove and proven treatment processes (i.e., reverse osmosis) come with significant limitations. Inland states in the US have effectively eliminated reverse osmosis as a feasible treatment process for WRRFs in the future, mainly due to their inability to permit brine management options.

This research will benefit WRF subscribers, utilities, and the water sector by identifying best practices for conducting regional salinity assessments, quantifying the relative contributions of common salt sources in watersheds, and proactively managing salinity as a growing water quality issue.

## Research Approach

This RFP is intentionally flexible in the research approach to encourage creativity and originality from proposers. Proposers should describe how they will conduct the research to meet the objectives listed above.

## Expected Deliverables

- A summary and comparative database on federal and state salt ion surface water standards and permit limits
- A review of successful case studies on source control of salts for various ions
- Guidelines on approaches for conducting watershed mass balances and source identifications
- Examples of communication material

Additional potential deliverables may include:

- Report focusing on a categorical overview of salinity source types, as well as a summary and resources from other national/regional entities
- Modelling results and options for defining TDS/salinity inputs and outputs
- Communication and educational materials for the water sector

Proposers are encouraged to partner with federal agencies such as the United States Geological Survey (USGS), River Basin Commissions and other research organizations/stakeholders to support the research's objectives/goal.

## Communication Plan

Please review WRF's [Project Deliverable Guidelines](#) for information on preparing a communication plan. Conference presentations, webcasts, peer-reviewed publication submissions, and other forms of project information dissemination are typically encouraged.

## Project Duration

The anticipated period of performance for this project is 24 months from the contract start date.

## References and Resources

The following list includes examples of research reports, tools, and other resources that may be helpful to proposers. It is not intended to be comprehensive, nor is it a required list for consideration.

- Carnegie Mellon University. 2010. Marcellus Shale Drilling. Accessed August 27, 2024. <https://www.cmu.edu/homepage/environment/2010/summer/marcellus-shale-drilling.shtml>
- Danboos, A., S. Sharil, F. M. Hamzah, A. Yafouz, Y. F. Huang, A. N. Ahmed, A. A. Ebraheem, M. Sherif, and A. El-Shafie. 2023. Water Budget-Salt Balance Model For Calculating Net Water Saving Considering Different Non-Conventional Water Resources in Agricultural Process. *Heliyon*, 9 (4): e15274.

- Fellet, M. 2014. Fracking Wastewater Could Encourage Formation Of Toxic Compounds During Drinking Water Disinfection. American Chemical Society: Chemical & Engineering News. Accessed August 27, 2024. <https://cen.acs.org/articles/92/web/2014/09/Fracking-Wastewater-Encourage-Formation-Toxic.html>
- Good, K. D., and J. M. VanBriesen. 2017. Power Plant Bromide Discharges and Downstream Drinking Water Systems in Pennsylvania. *Environmental Science & Technology*, 57 (20): 11829–11838.
- Kelly, V. R., M. A. Cunningham, N. Curri, S. E. Findlay, and S. M. Carroll. 2018. The Distribution of Road Salt in Private Drinking Water Wells in a Southeastern New York Suburban Township. *Journal of Environmental Quality*, 47 (3): 445–451.
- McCleskey, R. B., C. A. Cravotta III, M. P. Miller, F. Tillman, P. Stackelberg, K. J. Knierim, and D. R. Wise. 2023. Salinity and total dissolved solids measurements for natural waters: An overview and a new salinity method based on specific conductance and water type. *Applied Geochemistry*, 154: 105684.
- NHDES (New Hampshire Department of Environmental Services). 2021. Road Salt and Drinking Water. State of New Hampshire. Accessed August 27, 2024. <https://www.des.nh.gov/news-and-media/blog/road-salt-and-drinking-water>.
- Parker, K. M., T. Zeng, J. Harkness, A. Vengosh, and W. A. Mitch. 2014. Enhanced Formation of Disinfection Byproducts in Shale Gas Wastewater-Impacted Drinking Water Supplies. *Environmental Science & Technology*, 48 (19): 11161–11169.
- Rosenberg, E., J. Wietgreffe, M. Pazahanick, M. Sadler, and T. Walker. 2022. Addressing Increasing TDS & nbDON at Water Resource Reclamation Facilities. Hazen & Sawyer. Accessed August 27, 2024. <https://www.hazenandsawyer.com/articles/addressing-increasing-tds-nbdon-at-water-resource-reclamation-facilities>
- USEPA (United States Environmental Protection Agency). 2022. EPA Researching the Impacts of Freshwater Salinization Syndrome. Accessed August 27, 2024. <https://www.epa.gov/sciencematters/epa-researching-impacts-freshwater-salinization-syndrome>

### **Proposal Evaluation Criteria**

The following criteria will be used to evaluate proposals:

- Understanding the Problem and Responsiveness to RFP (maximum 20 points)
- Technical and Scientific Merit (maximum 30 points)
- Qualifications, Capabilities, and Management (maximum 15 points)
- Communication Plan, Deliverables, and Applicability (maximum 20 points)
- Budget and Schedule (maximum 15 points)

## **PROPOSAL PREPARATION INSTRUCTIONS**

Proposals submitted in response to this RFP must be prepared in accordance with WRF's [Guidelines for Research Priority Program Proposals](#) and [Instructions for Budget Preparation](#). These guidelines contain instructions for the technical aspects, financial statements, indirect costs, and administrative requirements that the applicant must follow when preparing a proposal.

Proposals that include the production of web- or software-based tools, such as websites, Excel spreadsheets, Access databases, etc., must follow the criteria outlined for web tools presented in the [Technology Deliverables Guidance](#).

### **Eligibility to Submit Proposals**

Proposals will be accepted from both U.S.-based and non-U.S.-based entities, including educational institutions, research organizations, governmental agencies, and consultants or other for-profit entities.

WRF's Board of Directors has established a [Timeliness Policy](#) that addresses researcher adherence to the project schedule. Researchers who are late on any ongoing WRF-sponsored studies without approved no-cost extensions are not eligible to be named participants in any proposals. Direct any questions about eligibility to the WRF project contact listed at the top of this RFP.

### **Administrative, Cost, and Audit Standards**

WRF's research program standards for administrative, cost, and audit compliance are based upon, and comply with, Office of Management and Budget (OMB) Uniform Grants Guidance (UGG), 2 CFR Part 200 Uniform Administrative Requirements, Cost Principles, and Audit Requirements for Federal Awards, and 48 CFR 31.2 Contracts with Commercial Organizations. These standards are referenced in WRF's [Guidelines for Research Priority Program Proposals](#) and include specific guidelines outlining the requirements for indirect cost negotiation agreements, financial statements, and the Statement of Direct Labor, Fringe Benefits, and General Overhead. Inclusion of indirect costs must be substantiated by a negotiated agreement or appropriate Statement of Direct Labor, Fringe Benefits, and General Overhead. Well in advance of preparing the proposal, your research and financial staff should review the detailed instructions included in WRF's [Guidelines for Research Priority Program Proposals](#) and consult the [Instructions for Budget Preparation](#).

### **Budget and Funding Information**

The maximum funding available from WRF for this project is \$150,000. The applicant must contribute additional resources equivalent to at least 33% of the project award. For example, if an applicant requests \$100,000 from WRF, an additional \$33,000 or more must be contributed by the applicant. Acceptable forms of applicant contribution include cost share, applicant in-kind, or third-party in-kind that comply with 2 CFR Part 200.306 cost sharing or matching. The applicant may elect to contribute more than 33% to the project, but the maximum WRF funding

available remains fixed at \$150,000. Proposals that do not meet the minimum 33% of the project award will not be accepted. Consult the [Instructions for Budget Preparation](#) for more information and definitions of terms.

### **Period of Performance**

It is WRF's policy to negotiate a reasonable schedule for each research project. Once this schedule is established, WRF and its sub-recipients have a contractual obligation to adhere to the agreed-upon schedule. Under WRF's [No-Cost Extension Policy](#), a project schedule cannot be extended more than nine months beyond the original contracted schedule, regardless of the number of extensions granted.

### **Utility and Organization Participation**

WRF encourages participation from water utilities and other organizations in WRF research. Participation can occur in a variety of ways, including direct participation, in-kind contributions, or in-kind services. To facilitate their participation, WRF has provided contact information, on the last page of this RFP, of utilities and other organizations that have indicated an interest in this research. Proposers are responsible for negotiating utility and organization participation in their particular proposals. The listed utilities and organizations are under no obligation to participate, and the proposer is not obligated to include them in their particular proposal.

### **Application Procedure and Deadline**

Proposals are accepted exclusively online in PDF format, and they must be fully submitted before 3:00 pm Mountain Time on Thursday, November 14, 2024.

The online proposal system allows submission of your documents until the date and time stated in this RFP. To avoid the risk of the system closing before you press the submit button, do not wait until the last minute to complete your submission. Submit your proposal at <https://forms.waterrf.org/cbruck/rfp-5308>.

Questions to clarify the intent of this RFP and WRF's administrative, cost, and financial requirements may be addressed to the WRF project contact, Lola Olabode, MPH, BCES, [lolabode@waterrf.org](mailto:lolabode@waterrf.org). Questions related to proposal submittal through the online system may be addressed to Caroline Bruck at 303.347.6118 or [cbruck@waterrf.org](mailto:cbruck@waterrf.org).

## ***Utility and Organization Participants***

The following utilities have indicated interest in possible participation in this research. This information is updated within 24 business hours after a utility or an interested organization submits a volunteer form, and this RFP will be re-posted with the new information. **(Depending on your settings, you may need to click refresh on your browser to load the latest file.)**

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